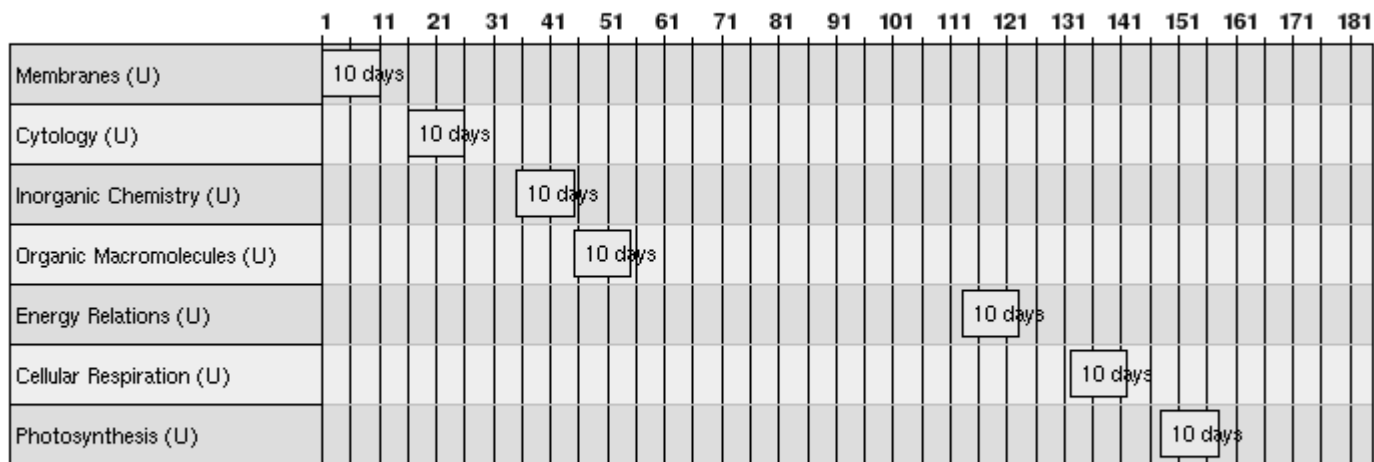


Folder: Science

Group/District: PENNSYLVANIA

Course Map Timeline Biology II



Essential (E)
 Important (I)
 Compact (C)
 Unranked (U)

Topic: Cellular Respiration

Days: 10

Subject(s):

Grade(s):

Key Learning:

The production of ATP through cellular respiration couples exergonic reactions to endergonic reactions in living things.



Unit Essential Question(s):

How are molecules of glucose oxidized to produce ATP?



<p>Concept: glycolysis and fermentation</p>	<p>Concept: kreb's cycle</p>	<p>Concept: electron transport</p>
<p>Lesson Essential Question(s): Why is pyruvic acid considered the hub of respiration? (A)</p>	<p>Lesson Essential Question(s): Why is the Kreb's cycle considered the energy wheel of respiration? (A)</p>	<p>Lesson Essential Question(s): What is the chemosmotic theory of cellular respiration? (A)</p>
<p>Vocabulary: glycolysis, fermentation, oxidative phosphorylation, substrate level phosphorylation</p>	<p>Vocabulary: acetyl coA, NAD+, FAD</p>	<p>Vocabulary: cytochrome, thermogenesis, hexose monophosphate shunt, Beta oxidation</p>

Additional Information:

Attached Document(s):

Vocab Report for Topic: Cellular Respiration

Days: 10

Subject(s):

Grade(s):

Concept: glycolysis and fermentation

glycolysis -
fermentation -
oxidative phosphorylation -
substrate level phosphorylation -

Concept: kreb's cycle

acetyl coA -
NAD+ -
FAD -

Concept: electron transport

cytochrome -
thermogenesis -
hexose monophosphate shunt -
Beta oxidation -

Topic: Cytology

Days: 10

Subject(s): Science

Grade(s): 11th, 12th

Key Learning:
The cell is the basic unit structure of all living things.



Unit Essential Question(s):

If all living organisms are comprised of cells, what causes the diversity in structure and function relationships within an organisms?

<p>Concept: cell types S11.B.1.1.2, S11.B.1.1.1, S11.A.1.1.4, S11.A.1.1.5, S11.A.1.1.1, S11.A.2.2.2, S11.B.2.1.1</p>	<p>Concept: viruses S11.A.1.1.4, S11.A.1.1.5, S11.A.2.2.2, S11.A.2.2.1</p>	<p>Concept: cellular organelles S11.B.1.1.3, S11.B.1.1.1, S11.A.1.1.5, S11.A.3.2.3, S11.A.1.1.1, S11.A.2.2.2, S11.B.2.1.1</p>
<p>Lesson Essential Question(s): What differences exist between prokaryotic and eukaryotic cells? (A)</p>	<p>Lesson Essential Question(s): How are viruses types distinguished from each other? (A)</p>	<p>Lesson Essential Question(s): Within a cell, what are the specialized components and their specific function? (A)</p>
<p>Vocabulary: conjugation, transformaton, plasmids, prokaryotic, eukaryotic, archaeobacteria</p>	<p>Vocabulary: capsid, nucleic acid, lytic, lysogenic</p>	<p>Vocabulary: Golgi apparatus, endoplasmic reticulum, nucleus, lysosomes, ribosomes, chloroplast, vacuole, mtiochondria</p>

Additional Information:

Attached Document(s):

Vocab Report for Topic: Cytology

Subject(s): Science

Days: 10

Grade(s): 11th, 12th

Concept: cell types

conjugation -
transformaton -
plasmids -
prokaryotic -
eukaryotic -
archaebacteria -

Concept: viruses

capsid -
nucleic acid -
lytic -
lysogenic -

Concept: cellular organelles

Golgi apparatus -
endoplasmic reticulum -
nucleus -
lysosomes -
ribosomes -
chloroplast -
vacuole -
mtiochondria -

Topic: Energy Relations
Subject(s): Science

Days: 10
Grade(s): 11th, 12th

Key Learning:
Endergonic reactions are driven by exergonic reactions in living systems.



Unit Essential Question(s):

How are endergonic reactions to proceed in living things?



Concept:
thermodynamic relationships
S11.C.1.1.2, S11.C.1.1.3, S11.C.1.1.6, S11.C.2.1.2, S11.C.2.1.3, S11.B.1.1.3, S11.B.1.1.2, S11.B.1.1.1

Concept:
enzymes
S11.C.1.1.2, S11.C.1.1.3, S11.C.2.1.1, S11.C.2.1.2, S11.C.2.1.3, S11.B.1.1.3, S11.B.1.1.2, S11.B.1.1.1



Lesson Essential Question(s):
How do the Thermodynamic Laws relate to energy relations in living things? (A)

Lesson Essential Question(s):
How do enzymes control metabolic processes n living things? (A)



Vocabulary:
entropy, enthalpy, exergonic, endergonic, bioenergetics

Vocabulary:
catalysts, allosteric, apoenzyme, holoenzyme,

Additional Information:

Attached Document(s):

Vocab Report for Topic: Energy Relations

Subject(s): Science

Days: 10

Grade(s): 11th, 12th

Concept: thermodynamic relationships

entropy -
enthalpy -
exergonic -
endergonic -
bioenergetics -

Concept: enzymes

catalysts -
allosteric -
apoenzyme -
holoenzyme -
-

Topic: Inorganic Chemistry

Days: 10

Subject(s): Science

Grade(s): 11th, 12th

Key Learning:

Review of the properties and behavior of inorganic compounds.



Unit Essential Question(s):

How does atomic structure determine the chemical characteristics of substances?



Concept:

Atomic Structure

[S11.C.1.1.1](#), [S11.C.1.1.2](#), [S11.C.1.1.3](#), [S11.C.1.1.4](#), [S11.C.1.1.6](#)



Lesson Essential Question(s):

How does atomic structure determine the chemical characteristics of an element? (A)



Vocabulary:

isotopes, ions, valence, oxidation, ionic, covalent

Concept:

Acids and Bases

[S11.C.1.1.1](#), [S11.C.1.1.2](#), [S11.C.1.1.3](#), [S11.C.1.1.4](#)



Lesson Essential Question(s):

How do we distinguish between acids and bases? (A)



Vocabulary:

proton, buffers, amphoteric

Concept:

Characteristics of Water

[S11.C.1.1.1](#), [S11.C.1.1.2](#), [S11.C.1.1.3](#), [S11.C.1.1.4](#)



Lesson Essential Question(s):

Why does water exhibit unique physical, chemical and mechanical characteristics? (A)

Why does water exhibit unique physical, chemical and mechanical characteristics? (ET)

How does the structure of water accounting for the physical, mechanical, and chemical properties of water? (A)



Vocabulary:

specific heat, cohesion, adhesion

Additional Information:

Attached Document(s):

Vocab Report for Topic: Inorganic Chemistry

Subject(s): Science

Days: 10

Grade(s): 11th, 12th

Concept:

Atomic Structure

isotopes -
ions -
valence -
oxidation -
ionic -
covalent -

Concept:

Acids and Bases

proton -
buffers -
amphoteric -

Concept:

Characteristics of Water

specific heat -
cohesion -
adhesion -

Topic: Membranes

Days: 10

Subject(s): Science

Grade(s): 11th, 12th

Key Learning:

Selectively permeable lipid bilayers found in all cells are composed primarily of proteins and lipids and are involved in a vast array of cellular processes.



Unit Essential Question(s):

What is the importance served by plasma membranes in living systems?



Concept:

plasma membranes

S11.B.1.1.1, S11.B.1.1.2, S11.B.1.1.3



Concept:

membrane transport

S11.B.1.1.3, S11.B.2.1.2



Concept:

Lesson Essential Question(s):
(A)



Lesson Essential Question(s):
How does the structure of the plasma membrane relate to metabolic needs of the cell? (A)



Lesson Essential Question(s):
How are materials transported through plasma membranes? (A)



Vocabulary:

hydrophobic, hydrophilic, glycocalyx, phospholipid, cholesterol

Vocabulary:
diffusion, osmosis, active transport, molal, molar, endocytosis, exocytosis, hypotonic, hypertonic



Additional Information:

Attached Document(s):

Vocab Report for Topic: Membranes

Subject(s): Science

Days: 10

Grade(s): 11th, 12th

Concept: plasma membranes

hydrophobic -
hydrophilic -
glycocalyx -
phospholipid -
cholesterol -

Concept: membrane transport

diffusion -
osmosis -
active transport -
molal -
molar -
endocytosis -
exocytosis -
hypotonic -
hypertonic -

Topic: Organic Macromolecules

Days: 10

Subject(s): Science

Grade(s): 11th, 12th

Key Learning:

Organization distinguishes organic molecules from inorganic molecules.



Unit Essential Question(s):

What molecular structures distinguish various organic molecules?

Concept: hydrocarbons S11.B.1.1.2, S11.B.1.1.3, S11.C.1.1.2, S11.C.1.1.3	Concept: functional groups S11.B.1.1.2, S11.B.1.1.3, S11.C.1.1.2, S11.C.1.1.3	Concept: carbohydrates S11.B.1.1.2, S11.C.1.1.2, S11.C.1.1.3
Lesson Essential Question(s): What is the basic hydrocarbon structure and identify the series to decane. (A)	Lesson Essential Question(s): What are the different types of organic functional groups and their associated characteristics? (A)	Lesson Essential Question(s): How do we distinguish between the different types carbohydrates and how their function of critical to all living things? (A)
Vocabulary: isomer, allotropic, alkyl groups, alkanes, alkenes, alkynes, arenes	Vocabulary: alcohol, ketone, ether, aldehyde, amide, amine, organic acid, ester	Vocabulary: monosaccharides, disaccharide, polysaccharide, oligosaccharide, glycogen
Concept: lipids S11.B.1.1.2, S11.B.1.1.3, S11.C.1.1.2, S11.C.1.1.3	Concept: proteins S11.B.1.1.2, S11.B.1.1.3, S11.B.2.2.1, S11.A.2.1.4, S11.A.2.1.5, S11.A.3.1.1, S11.A.3.1.3, S11.C.1.1.2, S11.C.1.1.3	Concept: nucleic acids S11.B.1.1.2, S11.B.1.1.3, S11.B.2.2.1, S11.A.2.1.4, S11.A.3.1.1, S11.A.3.1.3, S11.C.1.1.2, S11.C.1.1.3, S11.A.2.1.5
Lesson Essential Question(s): How do we distinguish between the different types lipids and how their function of critical to all living things? (A)	Lesson Essential Question(s): How do we distinguish between the different types proteins and how their function of critical to all living things? (A)	Lesson Essential Question(s): How do we distinguish between the different types nucleic acids and how their function of critical to all living things? (A) S11.A.2.1.5
Vocabulary: saponification, triglyceride, derived lipids, lipoproteins,	Vocabulary: keratin, peptide bond, denaturation, zwitterions, amino acid, collagen,	Vocabulary: transcription, translation, deamination, polymerase, nucleotide, redundancy, monocistronic

Additional Information:

Attached Document(s):

Vocab Report for Topic: Organic Macromolecules

Days: 10

Subject(s): Science

Grade(s): 11th, 12th

Concept: hydrocarbons

- isomer -
- allotropic -
- alkyl groups -
- alkanes -
- alkenes -
- alkynes -
- arenes -

Concept: functional groups

- alcohol -
- ketone -
- ether -
- aldehyde -
- amide -
- amine -
- organic acid -
- ester -

Concept: carbohydrates

- monosaccharides -
- disaccharide -
- polysaccharide -
- oligosaccharide -
- glycogen -

Concept:

lipids

- saponification -
- triglyceride -
- derived lipids -
- lipoproteins -
-

Concept: proteins

- keratin -
- peptide bond -
- denaturation -
- zwitterions -
- amino acid -

Vocab Report for Topic: Organic Macromolecules

Subject(s): Science

Days: 10

Grade(s): 11th, 12th

collagen -

-

Concept: nucleic acids

transcription -

translation -

deamination -

polymerase -

nucleotide -

redundancy -

monocistronic -

Topic: Photosynthesis

Days: 10

Subject(s): Science

Grade(s):

Key Learning:
Photosynthesis is the most fundamental biochemical process on the earth.



Unit Essential Question(s):

Why is photosynthesis considered the most important biological process on the earth?



Concept:

photosynthetic pigments

S11.C.1.1.2, S11.C.1.1.3, S11.C.2.1.1, S11.C.2.1.2, S11.B.1.1.2, S11.B.1.1.3



Concept:

light reactions

S11.C.1.1.2, S11.C.1.1.3, S11.C.2.1.1, S11.C.2.1.2, S11.B.1.1.2, S11.B.1.1.3



Concept:

dark reactions

S11.C.1.1.2, S11.C.1.1.3, S11.C.2.1.2, S11.B.1.1.2, S11.B.1.1.3



Lesson Essential Question(s):

What are the basic photosynthetic pigments and what is their importance to living systems? (A)



Lesson Essential Question(s):

How is photochemical energy captured by green plants? (A)



Lesson Essential Question(s):

How is carbon fixed into glucose? (A)



Vocabulary:

carotenoids, chlorophyll, action spectrum, absorption spectrum,

Vocabulary:

cyclic photophosphorylation, non cyclic photophosphorylation

Vocabulary:

Hatch-Slack pathway, Calvin Cycle

Additional Information:

Attached Document(s):

Vocab Report for Topic: Photosynthesis

Days: 10

Subject(s): Science

Grade(s):

Concept: photosynthetic pigments

carotenoids -
chlorophyll -
action spectrum -
absorption spectrum -
-

Concept: light reactions

cyclic photophosphorylation -
non cyclic photophosphorylation -

Concept: dark reactions

Hatch-Slack pathway -
Calvin Cycle -